C.U.SHAH UNIVERSITY Summer Examination-2022

Subject Name: Quantitative Optical Spectroscopic Methods-II							
Subject Code: 5SC04QSC1		Branch: M.Sc. (Chemistry)					
Semester: 4	Date: 04/05/2022	Time: 11:00 To 02:00	Marks: 70				

Instructions:

- (1) Use of Programmable calculator and any other electronic instrument is prohibited.
- (2) Instructions written on main answer book are strictly to be obeyed.
- (3) Draw neat diagrams and figures (if necessary) at right places.
- (4) Assume suitable data if needed.

SECTION – I

Q-1		Attempt the following questions	(07)
	a.	Define atomic absorption.	01
b.		Write temperature range for Acetylene-Air and Acetylene-Nitrous oxide flames.	01
	c.	Define nebulizer.	
	d.	What do you mean by atomizer?	01
	e.	Give the equation of Boltzmann distribution write the names of terms involved in the equation.	01
	f.	Write any two applications of ICP-AES.	01
	g.	Give the full form of ICP-OES and MES.	01
Q-2		Attempt all questions	(14)
	a.	Explain the Hollo Cathode Lamp and Electrodeless Discharge Lamp.	07
	b.	Discuss the Instrumentation of ICP-AES.	07
		OR	
Q-2		Attempt all questions	(14)
	a.	Write a note on flame atomizer.	07
	b.	Discuss the Interferences in AAS and ICP-AES.	07
Q-3		Attempt all questions	(14)
	a.	Write advantages, disadvantages and applications of AAS.	06
	b.	Write a note on nebulizer and spray chamberused in ICP-AES.	05
	c.	Give the difference between AAS and FES.	03



Q-3		Attempt all questions	(14)
¥۶	a.	Discuss the plasma torch and its types used in ICP-AES spectrometer.	06
	b.	Explain graphite tube atomizer.	04
	c.	Explain the CID detector used in ICP-AES spectrometer.	04
		SECTION – II	
Q-4		Attempt the following questions	(07)
	a.	An atomic fluorescence spectrometer is capable of measuring samples containing	01
		both hydride forming elements and Mercury at alevel.	
	b.	What is called Atomic fluorescence spectroscopy?	01
	c.	Define X-ray fluorescence	01
	d.	Give the equation showing relation between concentration and intensity for AFS.	01
	e.	Give any two types of fluorescence.	01
	f.	What is known as "lock and key" effectin AFS?	01
	g.	Write Bragg's law equation with name of all terms involved in the equation.	01
Q-5		Attempt all questions	(14)
	a.	Explain the Instrumentation of atomic fluorescence spectroscopy.	07
	b.	Give the advantages and disadvantages of XRF.	05
	c.	Draw labeled diagram of proportional counter detector use in WD x-ray spectrometer.	02
		OR	
Q-5		Attempt all questions	(14)
	a.	Discuss the instrumentation of X-Ray Fluorescence Spectrometry.	06
	b.	Give the applications of atomic fluorescence spectroscopy.	04
	c.	Write a note on the strength and limitations of XRF.	04
Q-6		Attempt all questions	(14)
	a.	Explain the instrumentation of Energy Dispersive XRF spectrometers.	05
	b.	Draw Jablonski diagram and explain various terms involved in the diagram.	05
	c.	Write the comparison of ED and WD X-ray spectrometry.	04
		OR	
Q-6		Attempt all questions	(14)
	a.	Explain the instrumentation of Wavelength Dispersive XRF spectrometers.	05
	b.	Give the applications of XRF.	05
	c.	Write a note on the micro XRF in detail.	04

